Mahishadal Raj College **Internal Assessment: 2018 Sub-Industrial Chemistry** Semester: I, Paper code: CC-1

F.M-10 Time -30 min.

(Answer five questions from the followings)

(5x2) = 10

- 1. Mention possible products of propane chlorination.
- 2. Describe synthesis of urea from CO2.
- 3. What is cellulose acetate? Give some applications.
- 4. What is Freon 12?
- 5. What do you mean by smock point and char value?
- 6. Distinguish between batch and continuous nitration.

Mahishadal Raj College **Internal Assessment: 2018 Sub-Industrial Chemistry** Semester: I, Paper code: CC-2

F.M-10 Time -30 min. (5x2) = 10

(Answer five questions from the followings)

- 1. Define the term "Smelting".
- 2. Explain the term "Setting of Cement" with appropriate equations
- 3. Write down the structures of Ortho and Meta silicates.
- 4. Define the term "Endurance".
- 5. Write down the composition of glass.
- 6. Give the chemical reactions take place during corrosion.

Year: 2nd, Paper code: III

F.M-10 Time -30 min. (Answer five questions from the followings) (5x2) = 10

- 1. What is mean by critical point of a drying rate curve?
- 2. Mention some advantages of cartridge filter?
- 3. What is catalyst poisoning? Give one example.
- 4. Define the term of Gold number.
- 5. What is katal?
- 6. What is relative volatility? Give its significant.

Mahishadal Raj College Internal Assessment: 2018 Sub-Industrial Chemistry Year: 2nd, Paper code: IV

- 1. State the uses of isopropanol.
- 2. What raw materials are used for manufacturing cumene industrially?
- 3. State the functions of column in gas chromatography
- 4. Explain the term surfactant.
- 5. How impurities of butadiene may be removed?
- 6. Explain the working principle of atomic spectrometry.

Year: 3rd, Paper code: VI F.M-10

Time -30 min.

(Answer five questions from the followings)

(5x2) = 10

- 1. Distinguish the following Offshore drilling and the drilling on land
- 2. Explain the terms Pour point
- 3. Which areas of the world has the most deposits of oil?
- 4. What is meant by API gravity?
- 5. Differentiate Octane number and cetane number.

Mahishadal Raj College Internal Assessment: 2019 Sub-Industrial Chemistry Semester: II, Paper code: CC-3

- 1. Define heat capacity of liquids.
- 2. State Hess's law.
- 3. Find out relation between percent saturation & relative saturation.
- 4. An aqueous solution of NaCl is prepared by dissolving 30 g NaCl in 100g water at 25°C. Find out the molality of the solution.
- 5. Explain the term in Antoine's equation.

Semester: II, Paper code: CC-4

F.M-10 Time -30 min. (Answer five questions from the followings) (5x2) = 10

- 1. Define the term "colloid" with example.
- 2. What are the differences between Heterogeneous & homogeneous catalysis?
- 3. State the Gibb's phase rule.
- 4. What is catalyst promoter. Give example.
- 5. What are micelles. Explain with an example.
- 6. Define the term enthalpy and internal energy.

Mahishadal Raj College Internal Assessment: 2019 Sub-Industrial Chemistry Year: 2nd, Paper code: III

F.M-10 Time -30 min.

(Answer the followings questions)

- 1. What is purge ratio?
- 2. State and explain Roults law?
- 3. What are filter aids?
- 4. Write the expression for Freundlich Isotherm.
- 5. What is aniline point?
- 6. What is the main different between a fan and a blower?

Mahishadal Raj College Internal Assessment: 2019 Sub-Industrial Chemistry Year: 2nd, Paper code: IV

F.M-10 (Answer five questions from the followings)

Time -30 min. (5x2) = 10

- 1. How errors of manometers can be rectified?
- 2. What raw materials are used for manufacturing cumene industrially?
- 3. What is BTX?
- 4. Write the composition of LNG.
- 5. What is AAS? Mention its uses.
- 6. Explain the term "Chemical shift".

Mahishadal Raj College Internal Assessment: 2019 Sub-Industrial Chemistry Year: 3rd, Paper code: VI

F.M-10 (Answer five questions from the followings)

Time -30 min. (5x2) = 10

- 1. Distinguish between Flash point and fire point.
- 2. Explain the terms OPEC.
- 3. What is meant by Pour point and pour point depressant
- 4. Discuss various steps used in the drilling an oil well..
- 5. What is visbreaking?
- 6. Name sulphur compounds present in petroleum.

Semester: I Paper code: CC-1

F.M-10 Time -30 min. (Answer five questions from the followings) (5x2) = 10

- 1. What are the major and minor inorganic constituent of coal ash?
- 2 What are the important product obtained from starch?
- 3. What are the component of a particular reforming unit?
- 4. Differentiate between thermal cracking and catalytic cracking with example.
- 5. What is nitrate ester? Mention some application.
- 6. What is hydroammonolysis? Why is it important?

Mahishadal Raj College Internal Assessment: 2019 Sub-Industrial Chemistry Semester: I Paper code: CC-2

- 1. What is roasting?
- 2. What is sludge disposal?
- 3. Write the composition of Portland cement.
- 4. Why silica is added to the roasted copper ores during smelting process?
- 5. What do you mean by anodising of alumina?
- 6. What do you mean by vulcanisation of rubber?

Semester: III Paper code: CC-5

F.M-10 (Answer five questions from the followings) Time -30 min. (5x2) = 10

- 1. Explain why change in dipole moment is essential for IR Adsorption.
- 2. Using 60 MHz instrument (NMR) the chemical shift of a photon was found to be $180~\mathrm{Hz}$. What would be the chemical shift if a $40~\mathrm{MHz}$ instrument is used?
- 3. Explain why ethanol is a good solvent for UV measurement but not for IR?
- 4. Write the full form of ISO and ISI standard.
- 5. What do you mean by Accuracy and precision?
- 6. State Lambert Beers law.

Mahishadal Raj College Internal Assessment: 2019 Sub-Industrial Chemistry Semester: III Paper code: CC-6

- 1. What are the common adsorbents used in TLC?
- 2. What is a chromatogram?
- 3. What are stationary phase and mobile phase in TLC?
- 4. Write down the basic principle of paper chromatography?
- 5. Mention few commonly used ion-exchanger.
- 6. What is a flame ionization detector?

Semester: III Paper code: CC-7

F.M-10 Time -30 min. (Answer five questions from the followings) (5x2) = 10

- 1. Explain the principle of thermal analysis.
- 2. What DSC can measure?
- 3. How does a glass electrode work?
- 4. Write down the basic principle of thermal mechanical analysis (TMA).
- 5. What are the advantages and application of gas sensing electrodes?

Mahishadal Raj College Internal Assessment: 2019 Sub-Industrial Chemistry Year-3rd, Paper code: VI

- 1. Distinguish between Viscosity and viscosity index.
- 2. Explain the term Doctoring.
- 3. What is meant by Octane number and cetane number
- 4. What is crude petroleum? Give its composition.
- 5. Discuss the role of estimation of different properties like specific heat.

Semester: II Paper code: CC-3

F.M-10 (Answer five questions from the followings) Time -30 min. (5x2) = 10

- 1. What is an ideal gas?
- 2. Define molarity & molality.
- 3. Distinguish between dry bulb temperature & wet bulb temperature.
- 4. Write down the general material balance equation for any unit operation.
- 5. Define the terms yields and selectivity.

Mahishadal Raj College Internal Assessment: 2020 Sub-Industrial Chemistry Semester: II Paper code: CC-4

F.M-10 (Answer five questions from the followings)

Time -30 min. (5x2) = 10

- 1. Define the term Gold number.
- 2. What is katal?
- 3. Write the expression for Freundlich Isotherm.
- 4. What do you mean by chain reaction? Explain with an example.
- 5. What is colloid? Give its partical size range?
- 6. What do you mean by extensive & intensive property.

Semester: IV Paper code: CC-8

F.M-10 Time -30 min. (Answer five questions from the followings) (5x2) = 10

- 1. Discuss froth flotation operation.
- 2. Distinguish extraction and leaching.
- 3. What is filter press?
- 4. Describe different sections of a packed tower.
- 5. State and explain Henry's law.

Mahishadal Raj College Internal Assessment: 2020 Sub-Industrial Chemistry Semester: IV Paper code: CC-9

- 1. Define heat transfer coefficient.
- 2. What is Fourier law of heat conduction?
- 3. Write down the significance of Bernaulli's equation.
- 4. Define the term Transmissivity.
- 5. Describe the operation of a vacuum pump.

Mahishadal Raj College Internal Assessment: 2020 Sub-Industrial Chemistry Semester: IV Paper code: CC-10

F.M-10 Time -30 min. (Answer five questions from the followings) (5x2) = 10

- 1. Give the principle of direct method and indirect method of liquid level measurement.
- 2. How is the measurement range limited in float type system?
- 3. What is the major difference of float level switch and displacer level switch measurement?
- 4. What is the principle of hydra step boiler drum?
- 5. What are the disadvantages of resistive method?

Mahishadal Raj College Internal Assessment: 2020 Sub-Industrial Chemistry Year-3rd, Paper code: VI

- 1. "Petroleum is recovered almost in the same way as underground water is obtained." Elaborate the statement
- 2. Explain different steps to recover petroleum.
- 3. Discuss the production procedure of petroleum coke from petroleum.
- 4. How can sulphur compounds be separated from petrolium?
- 5. Describe the manufacture of chloromethane from methane with essential reactions.

Mahishadal Raj College Internal Assessment: 2020 Sub-Industrial Chemistry Semester: I Paper code: CC-1

F.M-10 (Answer five questions from the followings) Time -30 min. (5x2) = 10

- 1. Discuss some advantages of continuous nitration over batch nitration.
- 2. How Benzene sulfonic acid is manufactured commercially?
- 3. Describe synthesis of urea from CO2.
- 4. What is vinyl acetate? How it is obtained?
- 5. Write down the chemical mechanism of amination by reduction
- 6. How oxalic Acid is manufactured?

Mahishadal Raj College Internal Assessment: 2020 Sub-Industrial Chemistry Semester: I Paper code: CC-2

F.M-10 (Answer five questions from the followings) Time -30 min. (5x2) = 10

- 1. Write down the structures of Ortho and Meta silicates
- 2. Define the term "Creeping"
- 3. What do you mean by "hard working and cold working"?
- 4. Explain the term "Galvanization" with its use .
- 5. Write the differences between thermoplastic and thermosetting materials with examples.
- 6. Write the composition and use of Duralumin.

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Semester: III Paper code: CC-5

F.M-10 (Answer five questions from the followings) Time -30 min. (5x2) = 10

- 1. Explain working principle of Flame photometer.
- 2. Acetone on boiling with barium hydroxide furnished two compounds A and B. How with the help of UV Spectroscopy A and B can be characterized?
- 3. Describe different absorption bands displayed by saturated aldehyde.
- 4. Write the full form of ISO and ISI standard.
- 5. In what respect FTIR is superior than IR spectrophotometer.

Mahishadal Raj College Internal Assessment: 2020 Sub-Industrial Chemistry Semester: III Paper code: CC-6

- 1. What is the role of detectors in chromatography?
- 2. Mention few solvents used in column chromatography.
- 3. Write down the principle of ion exchange chromatography.
- 4. What is the basic principle of paper chromatography?
- 5. What information you get from retardation factor value?

Semester: III Paper code: CC-7

F.M-10 Time -30 min. (Answer five questions from the followings) (5x2) = 10

- 1. What are the industiral applications of TMA.
- 2. Write down advantages of Gas sensing electrodes.
- 3. State the working principle of DSC.
- 4. Explain the principle of thermal analysis.
- 5. What are the applications of gas sensing electrodes?

Mahishadal Raj College Internal Assessment: 2020 Sub-Industrial Chemistry Semester: V Paper code: CC-11

- 1. Explain the mechanism of collision theory
- 2. What do you mean by Diffusion in liquid.
- 3. Define the homogeneous catalysis with examples.
- 4. Write down the Rungakutta method equation.
- 5. Explain the kinetics of chain polymerization reaction.

Semester: V Paper code: CC-12

F.M-10 Time -30 min. (Answer five questions from the followings) (5x2) = 10

- 1. What do you mean by knocking in petroleum?
- 2. Write about different types of drilling.
- 3. Why desalting of crude is necessary?
- 4. What are the impact of sulphur levels in gasoline & diesel?
- 5. Write the name of the feed stocks of cracker unit of a petrolium.

Mahishadal Raj College Internal Assessment: 2020 Sub-Industrial Chemistry Semester: V Paper code: DSE-1

- 1. What is triple superphosphate?
- 2. Give two examples of antimalarial drugs.
- 3. What is the process of hydrological cycle?
- 4. What is chlormycetin? What is chloromycetin used for?
- 5. How phosphoric acid can be synthesized?
- 6. Explain how dissolved oxygen affects the water quality.

Semester: V Paper code: DSE-2

F.M-10 Time -30 min. (Answer five questions from the followings) (5x2) = 10

- 1. Draw planes with Miller indices (111)
- 2. What do you mean by voids in solid state?
- 3. What do you mean by Frank-Condon principle. ?
- 4. Why 'Stokes' lines are more intense than 'anti-Stokes' lines.
- 5. What are the basic metallurgical principles?
- 6. Explain the reaction with equation involved in the Cyanide process of extraction of Silver.

Mahishadal Raj College **Internal Assessment: 2021 Sub-Industrial Chemistry** Paper code: CC-3 Semester: II,

Time -30 min. F.M-10 (5x2) = 10

(Answer five questions from the followings)

- 1. Define percent saturation & relative saturation
- 2. Calculate total heat requirement of conversions of 10 MT water at 30°C to steam at 100°C.
- 3. What do you mean by heat capacity of gases?
- 4. Write down material balance equation for absorption process.
- 5. Distinguish between partial pressure & vapor pressure.

Semester: II, Paper code: CC-4

F.M-10 Time -30 min. (Answer five questions from the followings) (5x2) = 10

- 1. What are the differences between order & molecularity of a reaction?
- 2. Write down the Vant Hoff equation.
- 3. Write a short note on aerosols.
- 4. Name some important catalyst that are industrially used?
- 5. What do you mean by associated colloids?
- 6. What are the differences between lyophobic and lyophilic sols?

Mahishadal Raj College Internal Assessment: 2021 Sub-Industrial Chemistry Semester: IV, Paper code: CC-8

- 1. What do you mean vacuum distillation?
- 2. Write down the main criteria of solution for gas absorption.
- 3. Write down the main criteria of solution for gas absorption.
- 4. Write down the basic characteristics of packed column.
- 5. Describe the basic principle Centrifugal Filtration.
- 6. Write down the short note on screw conveyor.

Semester: IV, Paper code: CC-9

F.M-10 Time -30 min. (Answer five questions from the followings) (5x2) = 10

- 1. Explain the principles of double pipe heat exchanger.
- 2. How forced convection is different from natural convection?
- 3. What is extended surface heat exchanger? How it affects heat transfer?
- 4. What is friction factor? How it is measured?
- 5. Can a compressor work as a vaccum pump?
- 6. A bus consists of diesel having centane number 45. What it actually means?

Mahishadal Raj College Internal Assessment: 2021 Sub-Industrial Chemistry Semester: IV, Paper code: CC-10

- 1. Write down the principle of Mercurry Barometer.
- 2. Write down advantages of Automatic control system.
- 3. What is Differential flow meter?
- 4. What do you mean by international temperature scale?
- 5 What are the disadvantages of Bourdon pressure gauge?
- 6. Discuss the main features of Open-loop control system.

Semester: VI, Paper code: CC-13

F.M-10 Time -30 min. (Answer five questions from the followings) (5x2) = 10

- 1. What are drug receptors?
- 2. How can a drug be targeted to a specific organ?
- 3. Discuss preparation and uses of glyceryl trinitrate.
- 4. Briefly explain the synthesis and uses of the drugs Diazepam
- 5. Describe fermentation process for the synthesis of lycine.

Mahishadal Raj College Internal Assessment: 2021 Sub-Industrial Chemistry Semester: VI, Paper code: CC-14

- 1. What do you mean by Graft copolymers? Give examples.
- 2. Write down the differences between Homopolymers and heteropolymers.
- 3. Define degree of crystallinity
- 4. Write down the mechanism of Addition polymerization.
- 5. Write a short note on emulsion polymerization.

Mahishadal Raj College Internal Assessment: 2021 Sub-Industrial Chemistry Semester: VI, Paper code: DSE-3

F.M-10 Time -30 min.

(Answer the followings questions)

- 1. What is pyrethrum and write the uses of it.
- 2. Write the chemical name (IUPAC) and structure of DDT.
- 3. What are the raw materials used for the production of D.D.T.?
- 4. Write the chemical name (IUPAC) and structure of Gammaxene.
- 5. How can carbaryal be synthesized from naphthaene?
- 6. What are the toxic side effects of Malathion?

Mahishadal Raj College Internal Assessment: 2021 Sub-Industrial Chemistry Semester: VI, Paper code: DSE-4

- 1. What are the advantages of perovskite solar cells?
- 2. Which type of solar cell is third generation solar cell?
- 3. Is passive or active solar heating better? Explain.
- 4. Is solar desalination efficient? Explain.
- 5. Explain the electrical properties of semiconductors.
- 6. Explain the working principle of perovskite solar cells.

Mahishadal Raj College Internal Assessment: 2021 Sub-Industrial Chemistry Semester: I Paper code: CC-1

F.M-10

Time -30 min. (5x2) = 10

(Answer five questions from the followings)

- 1 Write down the composition of natural gas.
- 2 What is peroxidation? Give one example.
- 3. What is chloral? Mention its use.
- 4. What is hydrolysis? Give one example hydrolysis with aqueous acid.
- 5. What are the materials used for the manufacture of furfural?
- 6. What do mean by carbonization of coal?

Mahishadal Raj College Internal Assessment: 2021 Sub-Industrial Chemistry Semester: I Paper code: CC-2

F.M-10 (Answer five questions from the followings)

Time -30 min. (5x2) = 10

- 1. What is synthetic rubber? Give examples.
- 2. Write down the composition of glass.
- 3. What are the different categories of solid wastes?
- 4. What do you mean by the term 'setting of cement'?
- 5. Explain the physical properties of glass.

Semester: III Paper code: CC-5

F.M-10 (Answer five questions from the followings) Time -30 min. (5x2) = 10

- 1. What do you mean by the term 'relative error'.
- 2. What is nebulization in Atomic Absorption Spectroscopy?
- 3. Calculate the mean and standard deviation of the following set of analytical result: 15.67, 15.69 and 16.03g.
- 4. Describe the basic principle of X-ray fluorescence spectroscopy.
- 5. Write down the applications of neutron diffraction analysis.

Mahishadal Raj College Internal Assessment: 2021 Sub-Industrial Chemistry Semester: III Paper code: CC-6

F.M-10 (Answer five questions from the followings) Time -30 min. (5x2) = 10

- 1. What are the limitations of paper chromatography technique?
- 2. Give examples of some commonly used ion exchanger.
- 3. Explain mobile phase and stationary phase with respect to gas chromatography.
- 4. What are the different applications of ion exchange chromatography?
- 5. What are the applications of Gas Chromatography?

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Semester: III Paper code: CC-7

F.M-10 (Answer five questions from the followings) Time -30 min. (5x2) = 10

- 1. What are the applications of gas sensing electrodes?
- 2. Write down the basic principle of thermal mechanical analysis.
- 3. What is amperometry?
- 4. Enumerate the relative advantages of differential thermal analysis (DTA).
- 5. Explain the relative disadvantages of differential scanning calorimetry (DSC).

Mahishadal Raj College Internal Assessment: 2021 Sub-Industrial Chemistry Semester: V Paper code: CC-11

- 1. Explain the mechanism of transition state theory.
- 2. What do you mean by chain polymerization reaction?
- 3. Define the heterogeneous catalysis with examples.
- 4. What is ionic chain reaction?
- 5 Define the explosive reaction with example.

Semester: V Paper code: CC-12

F.M-10 (Answer five questions from the followings) Time -30 min. (5x2) = 10

- 1. Name three types of crude based on the nature of compounds present.
- 2. Discuss the term Kerogen.
- 3. Distinguish the following Offshore drilling and the drilling on land.
- 4. State the basic difference between a drag reducing agent and viscosity reducer.
- 5. What is crude petroleum? Give its composition.

Mahishadal Raj College Internal Assessment: 2021 Sub-Industrial Chemistry Semester: V Paper code: DSE-1

F.M-10 (Answer five questions from the followings) Time -30 min. (5x2) = 10

- 1. What should be the criteria for designing a bioreactor?
- 2. How do you test for coliform in water?
- 3. What are greenhouse gases? Write two examples of greenhouse gases.
- 4. What is TOC and how it can be measured?
- 5. What are COD and BOD in water?

.

Semester: V Paper code: DSE-2

- 1. Symmetric stretch mode of vibration of carbon dioxide molecule is Raman active but infrared inactive. Explain.
- 2. What do you mean by Frank-Condon principle. ?
- 3. Why intensity or Rayleigh lines is more than Stoke's line?
- 4. What do you meant by allotropy of iron?
- 5. Give a clear distinction between 'metal sulphide smelting' and 'metal oxide smelting'
- 6. Melting point of alumina is very high but Al is obtained by the electrolysis of alumina at much lower temperature.

Semester: II, Paper code: CC-3

F.M-10 Time -30 min. (Answer five questions from the followings) (5x2) = 10

- 1. Define humidity & dew point.
- 2. How will you calculate average molecular weight of a gas mixture?
- 3. Make a material balance on distillation column.
- 4. How will you calculate average heat capacity of a gas mixture?
- 5. Define purge ratio & combined feed system.

Mahishadal Raj College Internal Assessment: 2022 Sub-Industrial Chemistry Semester: II, Paper code: CC-4

- 7. What are the differences of Adsorption & Absorption, explain with examples
- 8. Define the term Heterogeneous & homogeneous catalysis with example
- 9. Explain the term Catalyst "Promoter" & Catalyst "Poison" with Examples
- 10. Define the Langmuir and Freundlich isotherm with equation.
- 11. Define the term Autocatalysis and enzyme catalysts with examples.
- 12. Define the term "Colloid" and "Micell" with examples.

Mahishadal Raj College Internal Assessment: 2022 Sub-Industrial Chemistry Semester: IV, Paper code: CC-8

F.M-10 Time -30 min. (Answer five questions from the followings) (5x2) = 10

- 1. What is Rault's law & Henry's law?
- 2 What are azeotropes?
- 3 What are common packing material used in packed bed column?
- 4. Define rate of drying? What is bound moisture?
- 5. What is the basic difference between forward feed & backward feed multiple effect evaporator system?

Mahishadal Raj College Internal Assessment: 2022 Sub-Industrial Chemistry Semester: IV, Paper code: CC-9

- 1. Discuss Newtonian and Non-Newtonian fluid with example.
- 2. Define continuity equation and Berloni's equation with significance of each term
- 3. Name of some pumps which are used in Chemical Industries.
- 4. Write the statement of Fourier's law of heat conduction. Define the term convection.
- 5. Write principles of co-generation of boiler.
- 6. Draw the temperature profiles of parallel and counter current flow Heat exchangers.

Semester: IV, Paper code: CC-10

F.M-10 Time -30 min. (Answer five questions from the followings) (5x2) = 10

- 1. What is the principle of ultrasonic sensors?
- 2. Define Reynolds number?
- 3. What are the advantages of venturi flow meters?
- 4. Give the advantages of differential pressure method of level measurement.
- 5. What is the principle of nuclear radiation method?
- 6. How the mass flow rate is determined?

Mahishadal Raj College Internal Assessment: 2022 Sub-Industrial Chemistry Semester: VI, Paper code: CC-13

- 1. Give two examples of antifungal drug.
- 2. What is antileprotic? Give one example.
- 3. Write the chemical name and structure of cetirizine.
- 4. What are the side effects of ibuprofen?
- 5. What is antipyretic drug? When is it used?

Semester: VI, Paper code: CC-14

F.M-10 Time -30 min. (Answer five questions from the followings) (5x2) = 10

- 1. Define the term polyethylene and PVC with their chemical structure? Mention at least One industrial application of polyethylene and PVC.
- 2. Write the basic differences between homo-polymer and Co-polymer?
- 3. Define the term "RESIN". What do you mean by intrinsic viscosity of polymer?
- 4. What is basic differences between thermoplastics & Thermosetting?
- 5. Explain the role of Co-catalyst in polymerization reaction?
- 6. How to control density & molecular weight of polymer.

Mahishadal Raj College Internal Assessment: 2022 Sub-Industrial Chemistry Semester: VI, Paper code: DSE-3

F.M-10 Time -30 min.

(Answer the followings questions)

- 1. Write the uses with chemical structure of following pesticides: (2 x 3=6)
 - a) Malathion
 - b) Carbaryl
 - c) Carbofuran
- 2. Define term pesticides, what are the different types of pesticides on the basis of their chemical nature.

Semester: VI, Paper code: DSE-4

F.M-10 Time -30 min. (Answer five questions from the followings) (5x2) = 10

- 1. What is geothermal energy?
- 2. What do you mean by 1st generation solar cell?
- 3. Why is solar energy called renewable energy?
- 4. How will you measure solar radiation?
- 5. What are photovoltaic devices?

Mahishadal Raj College Internal Assessment: 2022 Sub-Industrial Chemistry Semester: I Paper code: CC-1

- 1. Differentiate between thermal cracking and catalytic cracking with example.
- 2. What is nitrate ester? Mention some application.
- 3. With example explain oxynitration.
- 4. Define halogenation with example.
- 5. What is hydroammonolysis? Why is it important?
- 6. What are the major and minor inorganic constituent of coal ash?

Mahishadal Raj College Internal Assessment: 2022 Sub-Industrial Chemistry Semester: I Paper code: CC-2

F.M-10 (Answer five questions from the followings) Time -30 min. (5x2) = 10

- 1. Explain why ceramic materials have high melting point?
- 2. What is Synthetic Rubber? Give one example.
- 3. Explain the term "Calcination"
- 4. Name of two important alloys of Copper and Aluminium.
- 5. Define the term "Vulcanization".
- 6. Write down the composition of "Gun Metal" with its uses.

Mahishadal Raj College Internal Assessment: 2022 Sub-Industrial Chemistry Semester: III Paper code: CC-5

- 6. What information can be obtained from IR Spectral measurement?
- 7. What do you mean by Accuracy and precision?
- 8. Explain why ethylene and ethyne unlike propene and propyne have no carbon-carbon multiple bond stretching vibration.
- 9. Explain why ethanol is a good solvent for UV measurement but not for IR?
- 10. Among acetone, dibromo-ethane and acetaldehyde which shows single peak in PMR Spectrum?
- 11. What do you mean by standard deviation?

Semester: III Paper code: CC-6

F.M-10 (Answer five questions from the followings) Time -30 min. (5x2) = 10

- 1. Write down the principle of chromatography.
- 2. What is Thin Layer Chromatography (TLC)?
- 3. What are normal phase and reverse phase chromatography?
- 4 Write down the principle of HPLC.
- 5. How GC is useful for qualitative and quantitative analysis?

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Mahishadal Raj College Internal Assessment: 2022 Sub-Industrial Chemistry Semester: III Paper code: CC-7

- 1. What is Amperometry?
- 2. Explain the principle behind the working of an ion selective electrode.
- 3. What is half wave potential?
- 4. Explain the basic principle of Voltammetry.
- 5. What is Dropping mercury electrode?
- 6. Write down the application of polarography?

Semester: V Paper code: CC-11

F.M-10 Time -30 min. (Answer five questions from the followings) (5x2) = 10

- 1. What is ionic chain reaction?
- 2. Define steady state approximation.
- 3. Define the heterogeneous catalysis with examples.
- 4. Write down the Lagrange's formula.
- 5. Explain the explosive reaction with example.

Mahishadal Raj College Internal Assessment: 2022 Sub-Industrial Chemistry Semester: V Paper code: CC-12

- 1. Discuss the role of different catalysts used in hydrocracking.
- 2. Differentiate Octane number and cetane number
- 3. State the basic difference of a drag reducing agent and viscosity reducer.
- 4. Distinguish Flash point and fire point.
- 5. What are important products obtained from crude petroleum?

Mahishadal Raj College Internal Assessment: 2022 Sub-Industrial Chemistry Semester: V Paper code: DSE-1

F.M-10 (Answer five questions from the followings) Time -30 min. (5x2) = 10

- 1. Name of different type fertilizer used for agricultural purposes
- 2. Write the major uses of acrylonitrile.
- 3. Explain the term "N-P-K ratio".
- 4. What is meant by "explosive limit".
- 5. Write the important use of Triple super Phosphate fertilizer .

Mahishadal Raj College Internal Assessment: 2022 Sub-Industrial Chemistry Semester: V Paper code: DSE-2

- 1. What is lambert-Beer law?
- 2. What is the basic principle of Thermogravimetric Analysis (TGA)?
- 3. Write down the basic principle of conductometric titration.
- 4. What are stationary and mobile phase?
- 5. What is the significance of Furrier Transform in IR?

Mahishadal Raj College **Internal Assessment: 2023 Sub-Industrial Chemistry** Semester: II, Paper code: CC-3

F.M-10 Time -30 min. (5x2) = 10

(Answer five questions from the followings)

1. Define standard heat of formation.

- 2. Distinguish between vapor pressure and partial pressure.
- 3. Write down Clapeyron equation and explain each term involved.
- 4. Define the terms limiting reactant and excess reactant.
- 5. Distinguish between dry bulb temperature & wet bulb temperature.
- 6. Discuss the utility of by-pass operation.

Mahishadal Raj College **Internal Assessment: 2023 Sub-Industrial Chemistry** Paper code: CC-4 Semester: II,

- 13. What do you mean by sol-gel process. Give one example.
- 14. State the Gibb's phase rule.
- 15. What do you mean by activation energy of a reaction?
- 16. Write down the differences between order and molecularity of a reaction.
- 17. What do you mean by Adsorption? Give one example
- 18. Explain the term extensive and intensive property. Give one example.

Semester: IV, Paper code: CC-8

F.M-10 Time -30 min. (Answer five questions from the followings) (5x2) = 10

- 1. What is filter press?
- 2 Distinguish extraction and leaching.
- 3 What is meant by fractional distillation.
- 4. Describe the operation of a pneumatic conveyer.
- 5. Describe crystallization process.

Mahishadal Raj College Internal Assessment: 2023 Sub-Industrial Chemistry Semester: IV, Paper code: CC-9

- 5. What is the basic difference between a blower and a compressor?
- 6. What is Fourier law of heat conduction?
- 7. Distinguish Newtonian and non-Newtonian fluid.
- 4. What is meant by octane number & cetane number?
- 5. State Kirchoff's law.
- 6. Distinguish natural and forced convection.

Semester: IV, Paper code: CC-10

F.M-10 Time -30 min. (Answer five questions from the followings) (5x2) = 10

- 1. State Bernoulli's theorem
- 2. Define stagnation point.
- 3. What is the major difference between float level switch and displacer level switch measurement?
- 4. What are the methods used in liquid level measurement in boiler drum?
- 5. What are the advantages of Hydra step gauge?
- 6. What are the advantages of orifice plate?

Mahishadal Raj College Internal Assessment: 2023 Sub-Industrial Chemistry Semester: VI, Paper code: CC-13

- 1. What does poison mean in drugs?
- 2. How can you reduce the side effects of cetrizine?
- 3. What is the synthesis of analgesic?
- 4. What are the symptoms of central nervous system drugs?
- 5. How is vitamin B-12 produced?
- 6. What is the solvent for streptomycin?
- 7. How is Lysine formed?

Mahishadal Raj College Internal Assessment: 2023 Sub-Industrial Chemistry Semester: VI, Paper code: CC-14

F.M-10 Time -30 min.

(Answer five questions from the followings) (5x2) = 10

- 1. Define the term polyethylene and PVC with their chemical structure?
- 2. Write the basic differences between homo-polymer and Co-polymer?
- 3. Explain the term UF and PF resin.
- 4. What is the basic differences between Crystallization & Crystallinity?
- 5. Write three basic differences between thermoplastics & Thermosetting.
- 6. What is the industrial application of Neoprene and Teflon?

Mahishadal Raj College Internal Assessment: 2023 Sub-Industrial Chemistry Semester: VI, Paper code: DSE-3

F.M-10 Time -30 min.

(Answer five questions from the followings) (5x2) = 10

- 1. What is limonene? Write the structure of it
- 2. Write the name of two natural and two synthetic pesticides.
- 3. Explain the termnatural and synthetic pesticides with example?
- 4. Write down the effect of pesticides on human beings.
- 5. Why rotenone is useful as an insecticide?
- 6. What is Azadirachtin? Write its chemical formula.

Semester: VI, Paper code: DSE-4

F.M-10 Time -30 min.

(Answer five questions from the followings) (5x2) = 10

- 1. Differentiate renewable & nonrenewable resources of energy?
- 2. What is solar PV system?
- 3. How solar energy can be utilized by converting it into thermal energy?
- 4. Differentiate irradiance and irradiation.
- 5. How can you measure solar radiation available?